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Cationic Peptides

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I. INTRODUCTION

It has been often observed that no new classes of antibiotics have been developed since the introduction of the first quinoline, nalidixic acid, in 1962. However, over the past decade scientists have discovered that one of nature's most persistent approaches against bacteria involves cationic peptides. For example, cationic peptides are the major mechanism of defense against microbes in insects and plants, a predominant local defense at host surfaces including the skins of amphibians and mucosa of mammals, and the major proteinaceous species of the dedicated antimicrobial defense cells of mammals—namely, neutrophils. These peptides have a variety of structures and functions that include antibacterial (Gram-positive and -negative), antifungal, antiviral, antiendotoxin, and anticancer activities. Thus, they present perhaps the most profound example of convergent evolution, in which a variety of different peptides have evolved to a common set of functions.

Cationic peptides were traditionally isolated from natural sources or synthesized by solid phase or solution phase chemistry. Moreover, they have recently been synthesized by recombinant DNA methods in bacteria (1), insect cells (2), and plants (3,4). The fact that cationic peptides are produced naturally by certain bacteria (e.g., see Chapter 17), as well as the newly discovered ability to synthesize virtually any peptide by recombinant means in bacteria (1), clearly merits the use of the term "antibiotic" for these compounds. Thus, cationic peptides represent not only the first new class of antibiotic in the past 30 years, but the world's first genetically engineering antibacterials.

II. OCCURRENCE OF CATIONIC PEPTIDES IN NATURE

Recently, we reviewed the natural cationic peptides in depth and identified 145 sequences that have been isolated from nature (5). Some of these are listed according to structural class in Tables 1 and 2. Cationic peptides are ubiquitous in nature; they have been identified in bacteria, fungi, plants, insects, crustaceans, amphibians, mammals, and humans.

Table 1 Examples of Cationic Peptides

Mammalian defensins (NP-1)	VVCACRRALCLPRERRAGFCRIRGRIHPLCCRR
β -Defensins (BNBD5)	EVVRNPQSCRWNMGVCIPISCPGNMRQIGTCFGRVPCCR
Insect defensins (Sapecin)	ATCDLLSGTGINHSAACAAHCLCRGNRGGYCNGKAVCVCRN
Tachyplesins (Tachyplesin)	RRWCFRVCYRGFCYRKCR
Thionins (Rabbitwood)	KSCCRNTWARNCYNVCRIPTISREICAKKCDCKIISSETTCPS-DYPK
Loops (Bactenecin)	RLCRIVVIRVCR
α -Helical (Cecropin A)	KWKFKKIEKMGRNIRDGIVKAGPAIEVIGSAKAI
Histidine-rich (Histadin 2)	MKFFVALILALMLSMTGADSHAKRHHGYKRKFHEKHSHRGY-RSNLYLDN
Tryptophan-rich (Indolicidin)	ILPWKPWWPWR
Proline-rich (Bac 5)	PFRPPIRPPPIRPPFYPPFRPPPIRPPPIFPPPIRPPFRPPLRFP

Generally speaking, these compounds provide relatively nonspecific defenses against microbes (Table 3). Even those compounds elicited by bacteria are known to function as bacteriocins that kill other bacteria, presumably as a mechanism of competition for an ecological niche.

A. Mammals

A variety of peptides are involved in the mammalian oxygen-independent antimicrobial defense mechanism. Defensins are a family of small (29–35 amino acids) arginine- and cysteine-rich peptides that have been isolated from a variety of mammals, including rats, rabbits, and humans (6,7). Six human defensins have been identified, four of which, human neutrophil peptides (HNP-1,2,3,4), were purified from polymorphonuclear leukocytes and two of which, human defensins (HD-5 and 6), have been detected in the intestinal Paneth cells by *in situ* hybridization. Mouse defensins, cryptidins, are also found in the Paneth cells of the small intestine. All six human defensins share sequence homology that includes six cysteine residues forming three disulfide bridges. This results in a β -pleated sheet secondary structure. Defensins, although capable of killing a wide range of bacteria, fungi, and viruses, are more active against Gram-positive than Gram-negative bacteria. In addition to their permeabilization of biological membranes, these peptides also exhibit chemotactic and endocrine regulatory activities (8).

Human defensins are synthesized as 94- to 100-amino-acid preprodefensins that contain a conserved 19-amino-acid N-terminal signal sequence that targets the peptide to the endoplasmic reticulum. This is followed by an anionic propiece, proposed to balance the cationic charge of the defensin (9).

A subset of defensins, the β -defensins, have been isolated from bovine neutrophils (10). A unique consensus sequence distinguishes these defensins from those described above, although both contain the characteristic three disulfide bridges. Tracheal antimicrobial peptide (TAP) isolated from the bovine respiratory tract also contains the triple disulfide motif but is specifically expressed in the respiratory tract (11). This peptide is active against Gram-negative and -positive bacteria and yeast.

A distinct family of peptides, termed the cathelicidins, has been isolated from mammalian neutrophils; these include the bovine peptides bactenecin 5 (Bac5) (12) and indolicidin (13), the porcine PR-39 (14), and the rabbit peptide CAP18 (15). These

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CAP18

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BAC5
INDOLICIDIN
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Figure 1 Alignment from Bac5 is denoted. Microbial domains are

peptides contain a protease inhibitor. The form is cleaved by and CAP18 is pre

B. Amphibians

Frog skin and frog bioactivity (18,19) was bombinin from related, related bombicocci (21).

A family of the African clawed bioactivity against protozoa (23,24).

These peptidic analogs has developed (22,26). The cl (PGLa, PGO, and the signal peptides

Cationic peptides to *lenta*—brevinin-1 both possess single been isolated from homology with otilabilizes membranes

C. Insects

Upon infection, in synthesized in the peptides include cecropins (34,35). Cecropins voltage-dependent the silk moth *Hyale* fly (sarcotoxin I) and peptides in that the

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Figure 1 Alignment of Bac5, indolicidin, and CAP18 proregion sequences. Sequence variation from Bac5 is denoted by underlining in the indolicidin and CAP18 prosequences. Cleaved antimicrobial domains are indicated by bold lettering. (Reproduced by permission of Zanetti et al. (16).)

peptides contain a highly conserved propeptide that is also homologous to the cysteine-rich protease inhibitor cathelin (16,17). The antimicrobial *N*-terminal region of these proforms is cleaved by elastase. An alignment of the deduced proforms of Bac5, indolicidin, and CAP18 is presented in Figure 1.

Frog skin and frog gastric mucosa are rich in peptides, and many of them have antimicrobial activity (18,19). One of the first antimicrobial peptides to be isolated from this source was bombinin from the species of frog *Bombina variegata* (20). This and subsequently isolated, related bombinins display a high level of antibacterial activity against staphylococci (21).

A family of amphipathic α -helical peptides, the magainins, has been identified in the African clawed frog (*Xenopus laevis*) (22). Magainin has a broad range of antimicrobial activity against Gram-positive and Gram-negative bacteria (23–25), fungi (24), and protozoa (23,24).

These peptides have been well characterized, and the analysis of many synthetic analogs has developed an understanding of the components required for biological activity (22,26). The cloning of the cDNA for magainin and other related amphibian peptides (PGLa, PGO, and xenopsin) has revealed that all are produced as precursor molecules, the signal peptides of which share considerable homology (23,27-30).

Cationic peptides have also been isolated from other species of frogs. For example, cationic peptides termed brevinins have been isolated from *Rana brevipoda* and *Rana esculenta*—brevinin-1 and brevinin-1E, respectively (31,32). These 24-amino-acid peptides both possess single C-terminal disulfide bonds and two prolines. Also, dermaseptin has been isolated from the South American frog *Phyllomedusa sauvagii*. This peptide has no homology with other amphibian peptides, but due to its amphipathic nature, it permeabilizes membranes in a similar fashion (19).

C. Insects

Upon infection, insects can produce a wide range of antimicrobial peptides, which are synthesized in the fat body and/or haemocytes and secreted into the haemolymph. Such peptides include cecropins (33), and defensin-like peptides such as sapecin and phormicin (34,35). Cecropins are highly amphipathic peptides containing 31–39 residues that form voltage-dependent channels in lipid membranes (36). They were initially isolated from the silk moth *Hyalophora cecropia* (37) and have subsequently been isolated from the flesh fly (sarcotoxin I) and *Drosophila* (38,39). Cecropins are distinct from other insect cationic peptides in that they contain no cysteine residues and fail to lyse eukaryotic cells (33),

Table 2 Structural Classes of Cationic Peptides

Class of peptide	Structural motifs	Sources	Examples
Mammalian defensins	3 β -strands 3 disulfides	Rat, rabbit, guinea pig, human neutrophils, rabbit alveolar macrophages, human, mouse Paneth cells	MCP, NP, HNP, GNCP, rat NP, cryptidins
β -defensins	3 disulfides β -stranded	Bovine neutrophils, trachea	TAP, BNBD
Insect defensins	3 disulfides 2 β -strands 1 α -helix	Dragonfly, blowfly, flesh fly	Phormicin, sapecin, sarcotoxin, royalisin
Tachyplesins	2 disulfides 2 β -strands	Pig leukocytes, crabs, amaranth plants, maize, turnip	Protegrins, polyphemusins, tachyplesins, Ac-AMP, 1AFP2, MBIP-1
Thionins	3 disulfides structure unknown	Maize, radish, rabbitwood, barley lead, rape, crambe	Mj-AMP1, trionin, crambin
α -Helical	amphipathic α -helix	Fruit fly, bees, frogs, toads, cattle	Bombolitin, bombinin, cecropins, magainins, melittin, dermaseptin
Loops	1 disulfide structures unknown	Bovine neutrophils, pit viper	Bactenecin, toxin 1
Histidine-rich	structures unknown	Primates, humans	Histadins
Tryptophan-rich	poly-L-proline II	Bovine neutrophils	Indolicidin
Proline-rich	poly-L-proline II	Fruit fly, honey bee, bovine neutrophils	Drosocin, abaecin, apidaecin, Bac5, Bac7

although they retain activity against Gram-negative and -positive bacteria in micromolar concentrations (37). Interestingly, cecropin-like peptides have now been isolated from the pig intestine (40). This latter peptide, cecropin P1, however, differs from the insect forms by not containing an amidated C-terminus and also in its tertiary structure (41).

Defensins have also been isolated from a variety of insect species (34,35). They share an array of six cysteine residues resulting in a tertiary structure containing three disulfide bridges but forming a structure that is distinct from mammalian defensins (42). These peptides instead share amino acid sequence homology and tertiary structure homology with royalisin from bees and charybdotoxin and defensin from scorpions (43-45).

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E. Microbes

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F. Plants

Thionins are spec example, barley p rich basic peptide

Sapaecin, an insect defensin isolated from the flesh fly *Sarcophaga peregrina*, consists of 40 amino acids including the conserved six cysteine residues (46) and is most active against Gram-positive bacteria (35).

A number of well-characterized novel antimicrobial peptides have been isolated from the honeybee (*Apis mellifera*). These include abaecin (47), the apidaecins (48), and hymenoptaecin (49). The apidaecins are a family of small (18 residues) proline-rich peptides, isolated from the haemolymph of the honeybee, which have activity against Gram-negative and plant-associated bacteria (48). Apidaecin precursors consist of cassettes of tandemly repeated sequences of the mature peptide preceded by dipeptides that are cleaved to produce the mature peptide (50). These precursors retain antibacterial activity, although the increasing dipeptide content reduces activity. Abaecin is a 34-amino-acid proline-rich peptide that has sequence homology with the apidaecins but has a different antibacterial spectrum and a delayed antibacterial effect (47). Hymenoptaecin, larger than the other bee-derived peptides at 93 residues, does not contain a high proline content nor the characteristic cysteine residues of defensins (49). This peptide has been shown to be active against the inner and outer membranes of Gram-negative bacteria (49).

D. Crustaceans

Tachyplesins (I, II, and III) are a class of antimicrobial peptides produced in the haemocytes of the horseshoe crabs *Tachyplesus tridentatus*, *Tachyplesus gigas*, and *Carcinoscorpius rotundicauda* (51–53). Two tachyplesin analogs, polyphemusin I and II, have also been isolated from the horseshoe crab *Limulus polyphemus* (52). Tachyplesins (17 amino acids) and polyphemusins (18 amino acids) contain four cysteine residues and subsequently form a rigid structure containing two disulfide linkages, which results in a stable structure resistant to low pH and high temperature (51). These peptides have activity against Gram-negative and -positive bacteria and fungi (51).

E. Microbes

Bacterial antibiotic proteins have been studied for many years since their initial discovery in the 1920s. Common among Gram-negative bacteria are the colicins; rarer are the peptide bacteriocins such as microcin B17. Among Gram-positive bacteria, peptide bacteriocins are the most commonly isolated. Of these, the cationic type A antibiotics such as nisin (54) and Pep 5 (55) isolated from *Lactococcus lactis* and *Staphylococcus epidermidis*, respectively, are the most characterized. These peptides contain such unusual amino acids as lanthionine, 3-methyllanthionine, and dehydrobutyrine (55,56). The mode of action of these peptides is by the formation of transient voltage-dependent pores in the cytoplasmic membrane (57,58). Such activity causes ion leakage from the cell and a breakdown in the electropotential across the cell membrane, resulting in death. Fungi, such as *Rhizomucor pusillus*, have also been shown to produce antibacterial peptides. *R. pusillus* produces sillucin, a defensin-like peptide active against Gram-positive bacteria (59).

F. Plants

Thionins are specific cationic peptides produced by plants in response to infection. For example, barley produces a leaf-specific thionin, BTH6 (60). In addition, other cysteine-rich basic peptides belonging to the superfamily of peptides that includes thionins and

mammalian and insect defensins have been isolated from the seeds of plants. These contain generally between 30 and 50 amino acids and between four and eight cysteine residues. Examples of such peptides include Ac-AMP (61) and Mj-AMP (62). Plant-derived peptides are most active against Gram-positive bacteria and fungi. IC_{50} (concentration inhibiting 50% of fungi) values for Ac-AMP against a broad spectrum of fungi are 2 to 10 $\mu\text{g/ml}$, and for Mj-AMP2, 0.5 to 20 $\mu\text{g/ml}$ against 13 plant pathogenic fungi (61,62).

III. ANTIMICROBIAL ACTIVITIES

A. Antibacterial Activities

It is a little difficult to assess the relative activities of cationic peptides compared with those of other antibacterial agents, for two basic reasons. The first is that many investigators working on cationic peptides utilize nonstandard assays. The appropriate method of measuring antibacterial activities is to determine a minimal inhibitory concentration (MIC) by either the broth dilution method, in which 10^3 – 10^4 bacteria are inoculated into a row of tubes containing serial twofold dilutions of antibiotics (63), or the agar dilution procedure, which involves incorporation of dilutions of the antibiotic into plates and subsequent spotting of 10^3 – 10^4 organisms onto the surfaces of the plates (64). In contrast, cationic peptides are often tested by measuring zones of clearance on plates spread with bacteria after inoculation of peptides into wells cut into the agar or onto paper discs (65), or by measuring the concentration of peptide killing 50% of bacteria in killing assays. The former method suffers from diffusion limitations of peptides, whereas the latter suffers from an uncertain relationship to MIC. Furthermore, the activities of cationic peptides tend to be reduced in media of high ionic strength or with high divalent cation concentrations.

With the foregoing general comments, cationic peptides have just moderate antibacterial activities compared with those of conventional antibiotics (Table 4). Nevertheless, cationic peptides do have certain highly desirable activities. First, they tend to have broad-spectrum activity that can encompass both Gram-negative and Gram-positive bacteria, although different cationic peptides often preferentially affect one or the other. Second, their activities do not appear to be compromised by resistance mechanisms that commonly appear in the clinic. Thus, common resistance mechanisms, such as methicillin resistance in *Staphylococcus aureus*, intrinsic antibiotic resistance in

Pseudomonas aeruginosa in *Enterobacteriaceae*, tetracycline efflux, and furthermore, they tend to be active against Gram-negative bacteria, such as *Burkholderia*.

Despite the fact that the MIC, in contrast to the peptide can cause a MIC, whereas other antibiotics are of order of magnitude.

The detailed mechanism of action of the most prominent cytoplasmic membrane targets advantage.

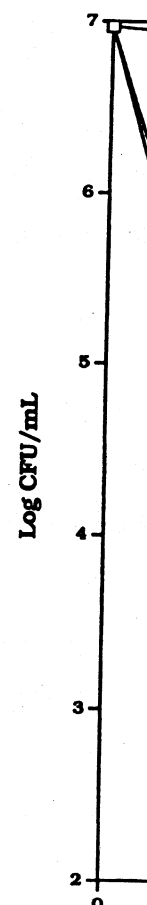


Figure 2 Killing of *Pseudomonas aeruginosa* by a cationic peptide.

Table 3 Roles of Cationic Peptides in Nature

Peptide	Host	Function	Mechanism
Pep5	<i>Staphylococcus epidermidis</i>	Antibacterial	Membrane disruption
Rs-AFP	Radish seeds	Antifungal	Cause hyperbranching and swelling of hyphae
Cecropins	Silk moth	Antibacterial	Membrane disruption
Tachyplesins	Horseshoe crab	Antibacterial, antifungal	Membrane disruption
Magainins	African clawed frog	Antibacterial, antifungal, antiprotozoal	Membrane disruption
Defensins	Human	Antibacterial, antifungal, antiviral	Membrane disruption

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sion in *Enterobacter cloacae* or plasmid-encoded TEM β -lactamase in *Escherichia coli*, and
tetracycline efflux in *E. coli*, have no effect on the MIC of cationic peptides (66). Fur-
thermore, they themselves do not tend to select resistant mutants, although some bacte-
ria, such as *Burkholderia cepacia* tend to be naturally resistant.

Despite their modest MICs, cationic peptides can kill bacteria potently at or around
the MIC, in contrast to most conventional antibiotics. Thus, a cecropin-melittin hybrid
peptide can cause 3–4 orders of magnitude of killing of *P. aeruginosa* in 20 min at the
MIC, whereas other potent antipseudomonal antibiotics generally cause less than one
order of magnitude of killing at the analogous concentration (Figure 2).

The detailed mechanism of action of cationic peptides is described in Section IV.C.
The most prominent effect on cells is the formation of channels in or disruptions of the
cytoplasmic membrane. Thus, these molecules appear to kill by a physical method that
takes advantage of the specific composition of bacterial membranes. In contrast, most

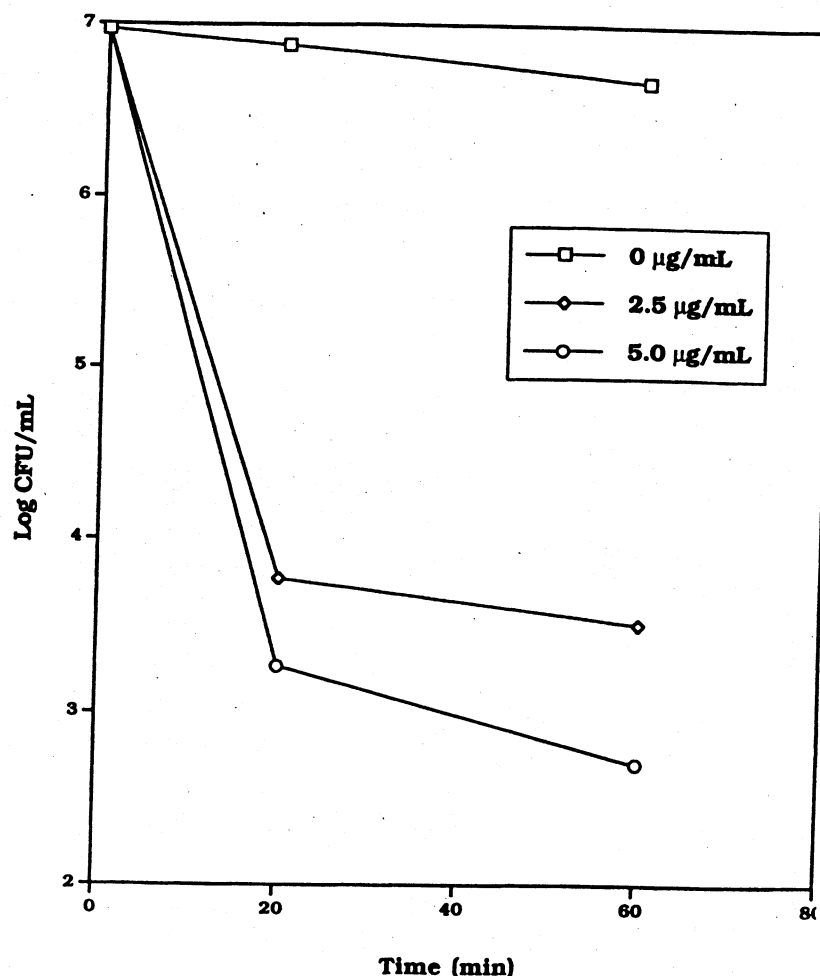


Figure 2 Killing of *P. aeruginosa* by the cationic peptide CEME. (From Ref. 67.)

conventional antibiotics are enzyme inhibitors that act on specific enzyme targets in bacteria (e.g., β -lactams acting on transpeptidases). This may explain many of the more desirable features of the cationic peptides, as described above, for example, (a) lack of resistance development, since it is difficult to fundamentally alter membrane composition, and (b) rapid killing, since the action is physical rather than catalytic.

Another feature of the mechanism of action that can be exploited is the ability of cationic peptides to break down the outer membrane barrier of Gram-negative bacteria. This barrier has been shown to limit the uptake of, and thus cellular susceptibility to, most conventional antibiotics, since its permeabilization by, for example, EDTA or specific mutations leads to reduced MICs for such antibiotics. In the same way, cationic peptides tend to be synergistic with certain conventional antibiotics, suggesting that they may be useful in the clinic in combination with such antibiotics. In keeping with these suggestions, Darveau et al. (68) demonstrated that magainin was synergistic with cefpirome in mouse protection experiments. The clinical implications of cationic peptides as antibiotic and antiendotoxin agents is discussed further in Sections II.B and IV.C.3.

B. Antiendotoxin Activities

Endotoxin is synonymous with lipopolysaccharide (LPS), a complex glycolipid that is an integral part of the outer membranes of Gram-negative bacteria (Figure 3). More specifically, endotoxin is the lipid A portion of LPS, which is the most membrane-proximal portion of the LPS making up the outer monolayer of the outer membrane. Endotoxin is a potent inducer of the cytokines interleukin 1 (IL1), tumor necrosis factor (TNF), and interleukin 8 (IL8). The presence of endotoxin in the body leads to a wide variety of physiological effects mediated in part by this vigorous cytokine response. These responses range from beneficial effects, such as fever response and tumor necrosis, to toxic effects,

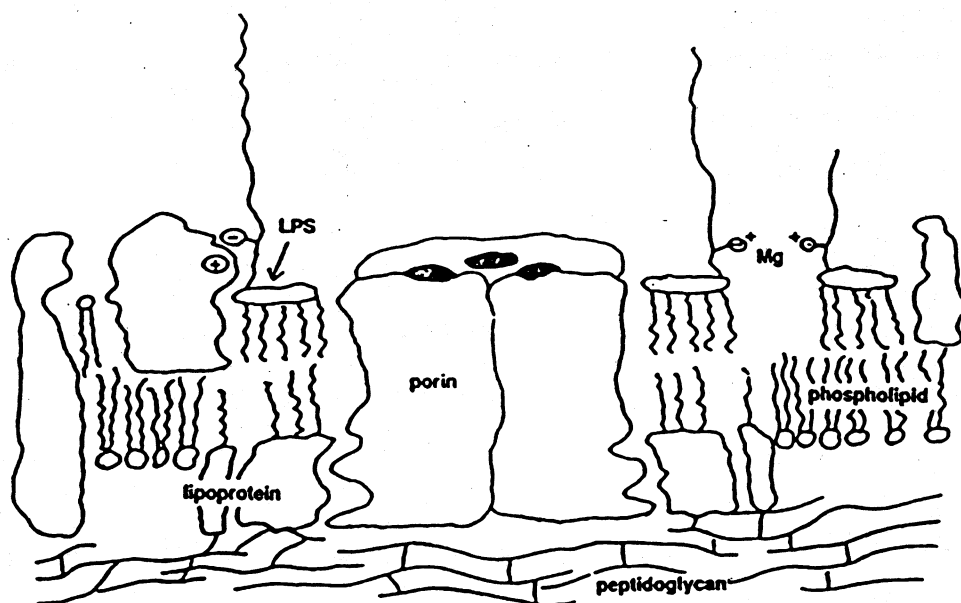


Figure 3 Schematic diagram of the outer membrane of a Gram-negative bacteria.

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LPS molecule, octasaccharide, charge that is neutral below, such anionic bacterial outer membrane peptides bind to bacteria higher than the cationic with macrophages frequently, cationic unlike other antishock, cationic p

C. Antifungal

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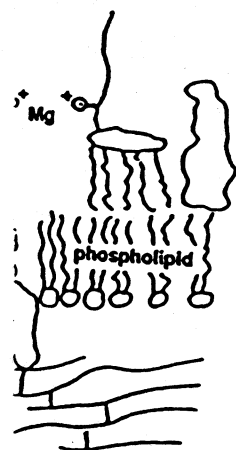
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including toxic or septic shock. It has been observed in the clinic that patients with Gram-negative bacterial blood infections can die even under circumstances where antibiotic treatment clears the infection. It is generally accepted that high endotoxin levels play an important role in determining lethality. One of the complicating factors is the tendency of antibiotics to promote LPS release both through lysis of bacteria and non-lytic mechanisms (69). Thus, it is of great interest to develop either antibiotics that do not enhance the release of LPS or treatments that neutralize released LPS.

LPS molecules contain several phosphate moieties in addition to the unique acidic octasaccharide, 2-keto-3-deoxyoctanate. Thus, they are strongly negatively charged, a charge that is neutralized in part by divalent cations such as Mg^{2+} and Ca^{2+} . As described below, such anionic residues are the site of initial interaction of cationic peptides with bacterial outer membranes. Indeed, it has been clearly demonstrated that cationic peptides bind to bacterial LPS with an affinity that is at least three orders of magnitude higher than the divalent cations (1,71,72). Such binding prevents LPS from interacting with macrophages to elicit a TNF response both in vitro and in vivo (72,73). Consequently, cationic peptides are protective in a mouse endotoxic shock model (74). Thus, unlike other antibiotics, which promote endotoxin release and consequent endotoxic shock, cationic peptides neutralize endotoxin and prevent endotoxic shock.

C. Antifungal Activity

Mammalian defensins kill *Candida albicans* within minutes in vitro (75). The action of such peptides involves four distinct steps: primary binding, postbinding events, permeabilization, and secondary binding to internal macromolecules (76). Other cationic peptides having antifungal activity include the tachyplesins and polyphemusins, which have MIC values against *C. albicans* of 3.1 and 6.3 $\mu g/ml$, respectively (52). Not surprisingly, those peptides isolated from plants have a broad range of antifungal activity against plant pathogenic fungi. For example, Rs-AFP, the antifungal peptide isolated from radish seed, has MIC values as low as 0.3 $\mu g/ml$ against certain plant pathogenic fungi (77). The plant peptides Mj-AMP and Ac-AMP cause a delay in growth of the fungal hyphae without changing mycelial morphology, whereas Rs-AFP causes a hyperbranching and swelling of the hyphae (61,62,77). However, these peptides show little or no activity against plant, insect, or human cells. Thionins cause a permeabilization of the plasmalemma around the hyphal tips (78).

D. Antiviral Activity

Several of the defensins have been found to neutralize herpes simplex virus (HSV) in tissue culture media (79). For example, rat NP-1 (50 $\mu g/ml$) caused direct viral neutralization, reducing HSV type 1 plaque forming units/ml by 90% in 60 min, and >99.9% of input viral titer was inactivated within 1 hr by 75 $\mu g/ml$ guinea pig defensin at 37°C (80).

E. Other Properties

In addition to the killing of the microorganisms described above, certain cationic peptides have been associated with the killing of parasites. Killing of *Giardia lamblia* has been demonstrated with indolicidin, cryptidins 2 and 3, and NP-2 (81). These peptides reduced the viability of the protozoal trophozoites by three orders of magnitude in 2 hr. The binding and lysis of the cells appears to involve charge interactions, as NaCl, Ca^{2+} , and Mg^{2+}

all inhibited killing. In addition, magainin analogs disrupted the morphological integrity and motility of several parasites, including *Entamoeba histolytica* and *Trypanosoma cruzi* (82). The latter was killed by 100 µg/ml of the magainin analog magainin B.

Other properties may exist for cationic peptides within the host. For example, sapaecin, the insect defensin, has been found to stimulate cell proliferation of *Sarcophaga* embryo cells. This perhaps indicates a dual role of antimicrobial agent and developmental hormone for this peptide in the flesh fly (83). Indeed, Magainin Pharmaceuticals, Inc., Plymouth Meeting, PA has claimed to have available cationic peptides that promote reepithelialization of damaged corneas.

Much work has been carried out on the potential for cationic peptides as anti-cancer agents. This has been most extensively studied with the magainins. An ovarian cancer murine model (84) showed the elimination of 99% of tumor cells after two injections of a magainin analog. In this study there was only mild damage caused to surrounding tissue, which indicates a higher susceptibility of malignant cells to these compounds.

F. Immunogenicity, Toxicity, and Stability

There has been no detailed examination of immunogenicity of cationic peptides. However, the general consensus in the field is that they are weakly immunogenic or nonimmunogenic. This could be due to clonal deletion during development, because of the importance of peptides in nonspecific host defenses at mucosal surfaces, and their secretion by neutrophils at sites of inflammation, resulting in their recognition as "self" antigens.

Although many cationic peptides are antimicrobial to some extent, their propensity to be toxic to mammalian cells varies greatly. For example, Schluesener et al. (85) found that although indolicidin and, to a lesser extent, battenecin are strongly cytotoxic to T lymphocytes, the defensins HNP-1, HNP-2, and HNP-3 did not affect the proliferation or viability of the T lymphocytes. On the other hand, the cationic peptides melittin and charyldotoxin are the potent toxins of bee and scorpion venom, respectively.

The issue of stability *in vivo* has not been addressed in detail. Clearly proteases, which are found in all body fluids, provide the potential for cleavage. One approach to overcoming such problems has involved synthesis of peptides with all D-amino acids; such peptides are not protease susceptible and often have equal activity to that of the L-form peptides (see Section IV.B).

IV. BIOCHEMICAL BASIS FOR ANTIBACTERIAL ACTIVITIES

A. Common Themes, Different Structures

All known cationic peptides share two properties: a high proportion of basic amino acids that are protonated and, thus, positively charged at neutral pH; and a high proportion of hydrophobic amino acids. These tend to distribute themselves through the three-dimensional structure of the peptide so as to create an amphipathic molecule having a hydrophilic, positively charged face and a hydrophobic face. However, the secondary and tertiary folding patterns of such peptides are quite diverse. For this reason, we consider the cationic peptides to be one of the finest arguments for convergent evolution, in which a variety of peptides have converged toward a common function, namely, defense against microbes.

The basic amino acids, lysine, arginine, and histidine, have a pI of 7-9. Lysine and arginine are protonated and positively charged at acidified sites. Histidine has a pI of 7 and is positively charged at acidified sites. In cationic peptides, the residues that can be positively charged are lysine, arginine, and histidine. These residues are often found in cationic peptides, such as cecropins or magainins, and are often found in pairs with aspartate and glutamate residues per peptide. The hydrophobic residues (hydrophilic) are often found in cationic peptides, such as cecropins or magainins, and are often found in pairs with lysine, arginine, and histidine residues, the cationic residues, due to their positive charge.

Despite the diversity of cationic peptides, they offer a range of structural classes. The defensins are the two-dimensional peptides with rather similar structures of triple-stranded and the entire structure variations, within the defensin family, and classes of defensins (Table 2). Although another class of cationic peptides, the insect defensins, the mammalian defensins, by NMR and show a helix, and a flexible structure. The characteristic evidence suggests (88). It has also been found in this case with

The second class of structures tend to be structured upon (89,90). The proline helix-bend-helix is a 11-14 amino acid dimensional NM cecropin-melittin cecropin P1, which is bounded by 2-4

Several cationic peptides have been demonstrated

The basic amino acids include arginine (pI = 10.8) and lysine (pI = 9.5), which are protonated and thus positively charged at pHs below their pI values. In addition, histidine has a pI of 7.6, rendering it only partially positively charged at neutral pH. However, at acidified sites in the body, including the upper gastrointestinal tract, the interior of phagocytic cell phagolysosomes, or some infection sites, histidine would be strongly positively charged. Nevertheless, with a few exceptions, histidines are relatively uncommon in cationic peptides. In contrast, cationic peptides usually contain four to nine positive residues that can comprise exclusively arginine (in defensins and thionins) or lysine (in cecropins or magainins) residues, or a mixture of the two. The acidic amino acids glutamate and aspartate are sometimes found in cationic peptides, but not more than two residues per peptide. The remaining (uncharged) residues often exclude several amino acids for a given peptide. Overall nonpolar (hydrophobic) residues usually exceed polar (hydrophilic) residues by a ratio of 2:1. Despite this high proportion of hydrophobic residues, the cationic peptides tend to be soluble in water, buffer, or acidified aqueous solutions due to their ability to fold and aggregate to mask their hydrophobic faces.

Despite their thematic similarities (i.e., their amphipathic nature), cationic peptides offer a range of secondary and tertiary folding patterns. The two most pronounced structural classes are the β -stranded and α -helical classes. The β -stranded class includes the defensins. The mammalian defensins have been crystallized (HNP-3) and studied by two-dimensional nuclear magnetic resonance (NMR) techniques (HNP-1, NP-2, NP-5) with rather similar results. They comprise two antiparallel β -strands with a short stretch of triple-stranded β -sheet (Figure 4). The β -strands are connected by short β -turn regions, and the entire structure is stabilized by three disulfide bridges. Despite some sequence variations, within a given class of defensins the positions of cysteines, the disulfide bonding patterns, and the positions of charged residues are strongly conserved. At least three classes of defensins exist, the mammalian defensins, β -defensins, and insect defensins (Table 2). Although not structurally well characterized, the plant thionins may make up another class of defensins. Only one other class has been examined structurally, namely, the insect defensins. These compounds have a different disulfide bonding array than do the mammalian defensins. The structure of the insect defensin, sapecin, has been defined by NMR and shown to contain two extended (β -stranded) regions, a short stretch of α -helix, and a flexible loop (87). However, the structurally characterized defensins all retain the characteristic hydrophobic surface and hydrophilic, positively charged surface. NMR evidence suggests that, in solution, defensins dimerize to mask the hydrophobic surface (88). It has also been demonstrated that tachyplesin adopts an amphipathic β -structure, in this case with two antiparallel β -strands stabilized by two disulfide bridges.

The second major structural class studied is the α -helical class. Interestingly, such structures tend to be rather disorganized in aqueous solution, but they become α -helical structured upon entering a membrane environment or exposure to nonpolar solvents (89,90). The predominant structures observed upon interaction with membranes are helix-bend-helix with a 9–16 amino acid amphipathic α -helix, a 2–4 residue bend, and a 11–14 amino acid amphipathic but more hydrophobic α -helix, as demonstrated by two-dimensional NMR of cecropins A and B, melittin, the magainins, and a synthetic cecropin-melittin hybrid (89,91–93). A small variation is provided by mammalian cecropin P1, which comprises an uninterrupted amphiphilic helix for 24 amino acids, bounded by 2–4 residues at the N- and C-termini.

Several cationic peptides are rich in proline, and specific peptides within this group have been demonstrated to adopt a poly-L-proline II helical structure (94–96). In the

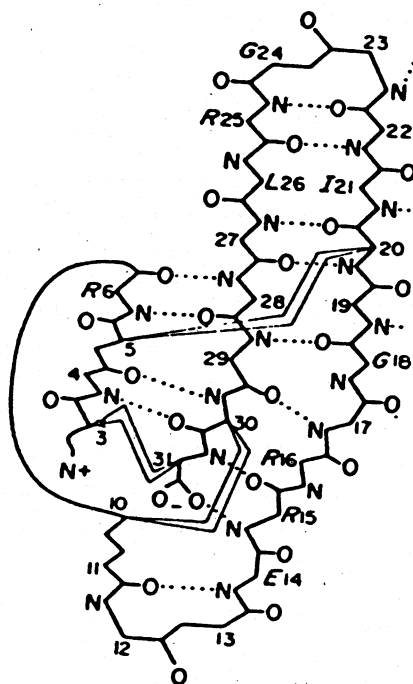


Figure 4 Model of the mammalian defensin structure. The triple-stranded antiparallel β -sheet structure of an HNP-3 monomer. The disulfide bonds are represented as "lightning bolts." Charged residues are indicated as R = arginine and E = glutamate. (Reproduced by copyright permission of Hill et al. (86). © American Association for the Advancement of Science.)

case of the proline/arginine-rich peptides, bactenecin 5 and PR-39, this structure is unaffected by the presence of lipid vesicles (95,96). However, in the case of the proline/tryptophan-rich peptide indolicidin, the assumption of this specific helical structure is greatly increased in the presence of negatively charged liposomes (94). Other peptides form loops due to single disulfide bonds, or they have extremely high histidine or tryptophan contents (Table 3). We anticipate that these will have a variety of structures.

B. Structure-Activity Relationships

The influence of substitutions or deletions of specific amino acids on the activity of the cationic peptides has been investigated in detail in several studies. The following principles seem to apply: (a) There is considerable specificity in how the changes in amino acid sequence influence activity (97,98). For example, introduction of a turn-promoting proline at positions 4 or 8 in the first α -helical segment of cecropins had a substantial effect on activity against *Micrococcus luteus*, lesser effects on activities against *Bacillus megaterium* and *P. aeruginosa*, and no effect on activity against *E. coli*. Furthermore, even conservative substitutions (changing selected amino acids to ones with similar physical properties) can substantially influence function. (b) For the α -helical peptides, changes that increase the tendency to form an α -helix in aqueous solution (i.e., prior to interacting with membranes) tend to increase activity (97-100). (c) There is no clear relationship

between the number of charges is important of the structured peptide. (e) Decreased activity correlates to some extent with absolute charge. For example, some cationic peptides are potent mediators of membrane lysis (unpublished data). (f) For the cationic peptides, activity is related to the net positive charge. (g) Finally, the overall structure of the peptide is important for activity. For example, some amino acids did not fit into the overall structure (102).

It must be stressed that the structure that will create the α -helicity favor an Ala-Ala-Lys) was a

C. Interactions with

The primary mechanism of action of channels in membranes is multistate channels. The channels are membrane target selective. The helical class, membrane target selectivity (i.e., the compound but a peptide the target lipid con

A secondary mechanism is unclear whether the channels or gross multistate mechanism is relevant. A nondefinitive experiment, the lysis of bacteria from the triggering of the ing how cationic pe

1. Model Systems

The process of interaction of the peptide dimers (e.g., defensins) in solution. The process is a cooperative, rapid response to the zeta potential of the membrane. The charges of the peptide respond to the zeta potential of the membrane. The charges of the peptide are posed of negatively charged residues and can dem

between the numbers of positive charges and activity, and the position of specific positive charges is important (98). (d) Enantiomers (i.e., all D-amino acids vs. all L-amino acids) of the structured peptides have equal activity (101), showing that chirality is not important. (e) Decreased ability of cationic peptides to bind to bacteria or to lyse liposomes correlates to some extent with decreased MIC (99). However, this correlation is not absolute. For example, one can design peptides that bind extremely well to Gram-negative bacteria but have little or no antibiotic activity (66; Hancock REW and Gough M, unpublished data). Furthermore, it must be noted that some cationic peptides that are potent mediators of liposomal lysis are potent toxins but relatively weak antibacterial agents. (f) For the disulfide-bonded peptides, reduction of the cysteine disulfides destroys activity. (g) Finally, there is no absolute correlation between peptide size and antibacterial potency. For example, reduction in size of cecropin-melittin hybrids from 26 to 14 amino acids did not influence activity so long as these compounds maintained an α -helical structure (102).

It must be stressed, however, that we do not at present have a set of design rules that will create the perfect peptide antibiotic. For example, although amphipathicity and α -helicity favor activity, a perfectly amphipathic α -helix (Lys-Ala-Ala-Lys-Ala-Ala-Ala-Lys) was a potent hemolysin.

C. Interactions with Membranes

The primary mechanism of action of cationic peptides is probably through the generation of channels in membranes. These can range from ordered channels through to so-called multistate channels. It must be stressed that we do not know in detail the basis for membrane target selectivity. For example, although they both fall into the amphipathic α -helical class, moth cecropins are strongly antibacterial and demonstrate minimal eukaryotic selectivity (i.e., toxicity), whereas melittin from bee venom is a weak antibacterial compound but a potent toxin. The primary basis for selectivity has been reported to be the target lipid composition (see below).

A secondary mechanism of action is a detergent-like effect (103). However, it is unclear whether this merely represents the cooperative accumulation of multistate channels or gross multimerization of cationic peptides in the membrane, and whether this mechanism is relevant to bacterial cell killing, since it has only been demonstrated in nondefinitive experiments in eukaryotic cell lines and model liposomes (103). In contrast, the lysis of bacteria—often at concentrations exceeding the MIC—probably arises from the triggering of autolytic enzymes (104). With these caveats, it is worth considering how cationic peptides interact with membranes.

1. Model Systems

The process of interaction of the peptides with lipid layers can be modeled as shown in Figure 5. The peptides initially present in solution as aggregates are present in the form of dimers (e.g., defensins) (88) and/or conformers (e.g., cecropins are relatively unstructured in solution). Interaction with the negatively charged head groups of lipids occurs in a cooperative, rapid process involving progressive binding and alignment of positive charges of the peptide with lipid head groups (105–107). The extent of binding corresponds to the zeta potential of the lipids involved, leading one to conclude that it is electrostatic in nature (106,107). Lipid composition is important: binding to liposomes composed of negatively charged lipids is extremely fast, but binding to zwitterionic lipids is slower and can demonstrate negative cooperativity (107).

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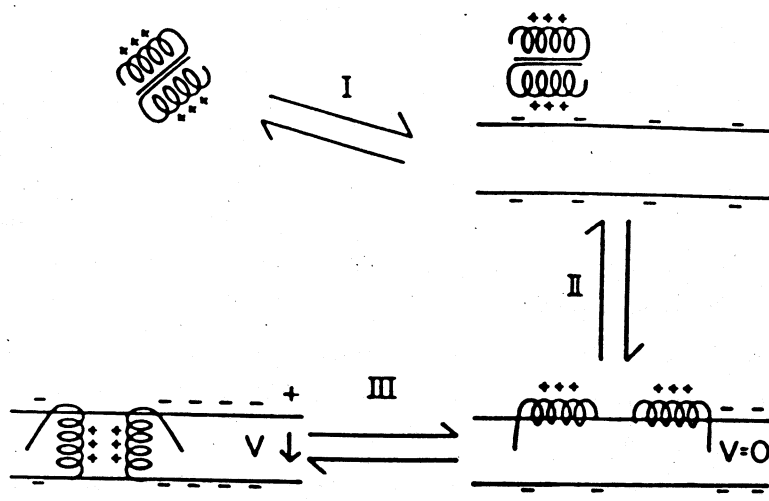


Figure 5 Tentative model for the interaction of cecropins with a lipid bilayer membrane. Aggregates adsorb to the bilayer-water interface by electrostatic forces (I). Only a dimer is sketched for the sake of simplicity, but larger aggregates are likely to occur. The next step (II) would be insertion of the hydrophobic segment into the membrane core. Upon application of voltage (positive on the side of the peptide addition), a major conformational rearrangement takes place (III), which results in channel formation. This rearrangement could be insertion of the positively charged amphipathic helix into the membrane or opening of preformed, closed channels. (Reproduced by copyright permission from Christensen et al. (36). © The National Academy of Sciences of the United States of America.)

It is uncertain whether at this point the permeability of the target lipid membrane changes. However, it seems credible that the phenomenon of leakiness (usually assessed by carboxyfluorescein leakage from liposomes) may occur in part at this stage. Furthermore, it is probable that the cationic peptide undergoes a change in conformation and aggregation state as a result of this interaction.

The next event is the insertion of the cationic peptide into the membrane. This occurs at a critical concentration of peptide that depends on the nature of the peptide and the target membrane, as well as on the fluidity of the membrane and the existence and size of the transmembrane electrical potential. Although insertion can occur into membranes with little or no transmembrane potential, it seems likely that the membrane potential of living cells (oriented interior negative) is always an important factor in peptide insertion. In addition, it has been demonstrated in planar lipid bilayer model membrane experiments (in which the "membrane potential" is provided as an applied voltage) that this potential must be oriented positive on the cis side (where the cationic peptides are added) and negative on the trans side of the membrane (toward which the cationic peptides move as they enter the membrane). This results in an observable increase in conductance as the peptides enter the membrane and form channels (36,94,108-111). Reversal of the voltage actually causes peptides to leave the membrane (36). It is relevant to the issue of toxicity that bacterial cytoplasmic membranes bear large transmembrane electrical potential gradients, $\Delta\psi$ (up to -140 mV), whereas eukaryotic membranes have gradients of only about -20 mV or less.

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2. Bacterial C

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The process of insertion causes changes in phase and/or motion of the lipids of the target membrane (112). However, the lipid composition can dramatically influence the possibility of insertion, and positively charged phospholipids and cholesterol decreased the formation of membrane channels by cecropin by 5- to 60-fold (36). Indeed, this may explain, in addition to the difference in $\Delta\psi$, the selectivity of cationic antibacterial peptides for bacteria over eukaryotes, since the former lack cholesterol, which is abundant in eukaryotic membranes, whereas anionic phosphatidyl glycerol and cardiolipin, major components of bacterial membranes, represent excellent target lipids.

The process of insertion can also cause a conformational change in the cationic peptides (e.g., from unstructured to α -helical), for example, with melittin (113) and magainins (90,114). In many cases, the peptides are thought to end up spanning the membrane bilayer (41,102,113) in multimeric complexes. Other peptides are too short to span the bilayer and presumably must form aggregates to permit transmembrane channel formation (102,114,115).

Generally speaking, cationic peptides form multistate channels, and planar lipid bilayer experiments demonstrate a substantial range of channel sizes, with single-channel conductances (which reflect size) varying from 10 to 2000 pS (94,108–110,116) and lifetimes ranging from milliseconds to seconds (110). This behavior is similar to that observed for alamethicin (117), for which it has been proposed that application of a voltage induces alamethicin monomers to span the membrane, and that these monomers associate and disassociate with various rate constants, leading to aggregates of different sizes and lifetimes. Each aggregate contains a variable number of monomers arrayed like staves of a barrel around a central axis, and oriented with the hydrophilic portion of the monomer facing inward toward the channel interior and the hydrophobic face adjacent to the membrane interior. Thus, the size of the aggregates would decree the size of the channel. These channels are water filled and tend to be weakly selective for chloride over sodium ions.

In specific instances, for example, the cecropins, the channels formed are more defined (118). In this case, the actual channel forming unit has been modeled at atomic-level resolution. Two arrangements of six dimers have been proposed to account for the two discrete conductance increments (0.4 and 1.9 nS) reported by Christensen et al. (36).

2. Bacterial Cytoplasmic Membranes

As discussed above, cationic peptides can form channels in model bilayers. Thus, it seems likely that their primary antibacterial action is to disrupt the integrity of bacterial cytoplasmic membranes. This would have the effect of permitting leakage of ions and small metabolites, and destroying the ability of bacteria to maintain a transmembrane proton gradient (proton-motive force) with consequent loss of ability to generate adenosine triphosphate and transport substrates (see Ref. 119 for review of cytoplasmic membranes).

Bacteria maintain, across their cytoplasmic membranes, a proton-motive force of approximately -170 mV (120), comprising an electrical potential gradient $\Delta\psi$ (oriented interior negative) and a proton chemical gradient ΔpH (oriented interior alkaline). Treatment of cells with any of several different cationic peptides (e.g., magainins, nisin, or Pep5) leads to destruction of $\Delta\psi$ at concentrations approaching the MIC, as assessed using the cationic lipid-soluble probe triphenyl phosphonium (121,122). Evidence suggests that this decrease of proton-motive force occurs as a sigmoidal function of peptide

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concentration, indicating that peptides act in a cooperative fashion on cytoplasmic membranes (121). Other data favoring the hypothesis that destruction of cytoplasmic membrane integrity is the primary basis of the activity of cationic peptides against bacteria include the demonstration that mastoporan and melittin cause K^+ leakage in Gram-positive bacteria (123).

3. Bacterial Outer Membranes

Only Gram-negative bacteria have outer membranes, and it is clear that the interaction of an antibiotic with outer membranes cannot directly lead to cell death. However, outer membranes are discussed separately here for two reasons. First, the cationic peptides include molecules that are rare among antibiotics in having better activities against Gram-negative than Gram-positive bacteria (normally, the influence of the outer membrane on penetration of antibiotics decreases activity). Second, the interaction of cationic peptides with the outer membranes of Gram-negative bacteria explains two of the pharmaceutically interesting properties of these molecules, namely, their "enhancer" and antiendotoxin properties.

Cationic peptides, like other polycationic antibiotics, traverse the outer membrane using a process termed self-promoted uptake (124); in contrast, small hydrophilic antibiotics such as β -lactams diffuse through the water-filled channels of porin proteins (125). Self-promoted uptake (Figure 6) involves the initial interaction of cationic peptides with the negatively charged, divalent-cation-binding sites of the surface glycolipid lipopolysaccharide (LPS). Since the cationic peptides have an affinity for LPS that is three orders of magnitude higher than the native divalent cations, Mg^{2+} or Ca^{2+} (70), they competitively displace these cations. This causes a distortion of outer membrane structure that has been visualized in the electron microscope as induction of outer membrane blebs (70), and a consequent permeabilization of the membrane to probe molecules, including lysozyme and the hydrophobic probe 1-N-phenyl-naphthylamine (71,126). By analogy with other polycations, this distortion of the membrane is proposed to lead to enhanced ability of the cationic peptide to promote its own uptake (hence the term self-promoted uptake). The basic features of this uptake system have been demonstrated for interaction of both the α -helical (66) and β -structured cationic peptides (70).

The ability of cationic peptides to act in synergy with certain classical antibiotics (68) can be explained by their ability to disrupt outer membrane integrity, promoting the uptake of antibiotics across this barrier. Interestingly, the most potent cationic peptides do not have this "enhancer" activity for most antibiotics, presumably since they kill cells

Table 4 MICs of Selected Cationic Peptides Compared with Conventional Antibiotics

Organism	MIC ^a (μ g/ml)				
	CP-29 ^b	CP-11c ^c	Gentamicin	Ceftazidime	Polymyxin
<i>Pseudomonas aeruginosa</i>	4	8	0.3	0.5	0.3
<i>Escherichia coli</i>	2	2	0.3	0.3	0.1
<i>Staphylococcus aureus</i>	16	8	2	2	>64
<i>Candida albicans</i>	32	8	>64	>64	>64

^a MIC values were determined by the broth dilution method (63).

^b CP-29 is a cecropin/mellitin hybrid cationic peptide (Hancock, Gough, and Farmer, unpublished data).

^c CP-11c is an extended helix cationic peptide (Falla, T., Hancock, R. E. W., unpublished data).



Figure 6 Schematic of Mg^{2+} cross-bridges, disruption of the outer

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The antiendotoxin mechanism. Endotoxin mentioned above, is of high affinity to induce TNF in macrophages, leading to mortality in galactosemia.

V. PRODUCTION

A. Natural Sources

As described in Section I, many of these sequences are extracted from natural sources. As described in Section I, many of these sequences are extracted from natural sources. As described in Section I, many of these sequences are extracted from natural sources.

B. Protein Chemistry

A very convenient method for the synthesis of cationic peptides is the solid-phase method.

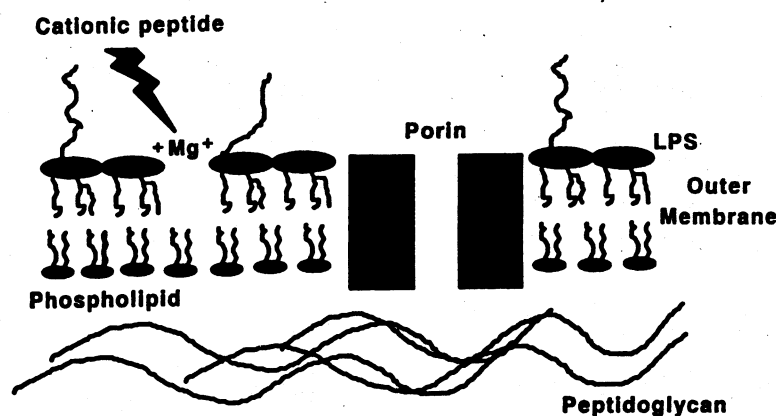


Figure 6 Schematic diagram of the self-promoted uptake model. Cationic antibiotics disrupt the Mg^{2+} cross-bridges, displace the Mg^{2+} ions, and cause perturbation of the lipid bilayer. Further disruption of the outer membrane results in uptake of normally excluded compounds.

at concentrations equal to their permeabilizing concentrations (66). This is analogous to the situation for the polycationic antibiotic polymyxin B, which is not an enhancer, whereas its deacylated derivative PMBN (which interacts weakly with cytoplasmic membranes but strongly with outer membranes) is a potent enhancer of antibiotic activity (127).

The antiendotoxin activity of cationic peptides is also related to the above uptake mechanism. Endotoxin is in fact LPS, or more precisely the lipid A portion of LPS. As mentioned above, cationic peptides bind to polyanionic LPS (70,128,129). The binding is of high affinity and cooperative (70). This binding can neutralize the ability of LPS to induce TNF in macrophage cell lines or in a murine model, and it reduces endotoxin mortality in galactosamine-sensitized mice (130,131).

V. PRODUCTION METHODS

A. Natural Sources

As described in Section II, cationic peptides are very widely distributed in nature. Recovery from these sources involves a wide range of methods. One effective procedure is extraction with 30% acetic acid, which tends to solubilize cationic peptides and precipitate many globular proteins. This is usually followed by a variety of chromatographic procedures often including reverse phase HPLC or FPLC as the final step in purification. However, purification from natural sources is rarely a practical alternative for commercial purposes, since yields tend to be relatively low. For example, a single rabbit will permit the recovery of only 200 mg of rabbit defensins. The one exception is the production of cationic lantibiotic peptides such as nisin from bacteria and commercial production of nisin by fermentation of *Lactococcus lactis* (see Chapter 15).

B. Protein Chemical

A very convenient laboratory-scale procedure for making peptides is the use of automated peptide synthesizers using t-boc or f-moc chemistry (132). However, the expense

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of reagents and the limited capacities of these automated synthesizers has limited the scale and, thus, the industrial relevance of this method. An alternative is provided by solution phase chemistry, which is, unfortunately, less conducive to automation.

C. Recombinant Procedures

One potential advantage of the peptide nature of cationic peptides is their potential ability to be synthesized recombinantly, since they can be directly encoded by DNA. There are certainly some limitations to this, since nonnatural amino acids (e.g., in the antibiotics) and many modifications (e.g., carboxyl-terminal amidation) are difficult to introduce recombinantly. As with the protein chemical procedures, however, one is not limited to "natural" cationic peptides.

The first attempt to synthesize cationic peptides recombinantly appears in the patent literature (133). A sequence encoding cecropin was fused to a portion of the *araC* gene of *E. coli*. Although few details were provided, it is clear that this method was not optimized: although cecropin could be manufactured recombinantly, it had poor potency (i.e., 5 µg of recombinant cecropin gave a clearing zone diameter against *E. coli* of 5 mm, whereas 1 µg of authentic cecropin gave a clearing diameter of 7 mm). Furthermore, the patent claimed that virtually any fusion partner would work to support the production of cecropins, whereas it is now evident that this is not the case. Subsequently, it was demonstrated that cecropin could be synthesized as a fusion with a protein A-like, IgG-binding domain, using baculovirus vector in an insect cell line (2). After affinity purification of the fusion protein and cleavage of cecropin from its carrier using cyanogen bromide, the cecropin could be recovered in its amidated form. The yields were 600 µg/ml of haemolymph, of which 70% was amidated, indicating that this method may be cost prohibitive given the expense associated with animal cell culture.

Piers et al. (1) have developed a procedure for the synthesis of cationic peptides in bacteria. The main feature is inclusion of an anionic stabilizing fragment in the fusion protein to counteract the cationic peptide portion. This anionic fragment could be the carrier protein itself if the fusion protein was expressed in *S. aureus*, but for expression in *E. coli* an extra anionic sequence equivalent to the pre pro sequence from the gene for human defensin (which sequence stabilizes defensin during its synthesis in human cells; 134) is needed. Additional elements were the inclusion of a methionine residue immediately adjacent to the cationic peptide sequence, to permit removal of the cationic peptide by CNBr cleavage, and a carrier sequence that, when desired, could be tailored to enhance affinity purification of the resultant fusion. The general nature of the fusion protein is demonstrated in Figure 7. We believe this system offers significant advantages in the production of cationic peptides. An interesting side note is that the successful production of antibacterial cationic peptides by molecular genetic means makes these cationic peptides the first recombinant antibiotics.

VI. AN EXCITING FUTURE

Cationic peptides do not have as potent activities against the most susceptible bacteria as do other antibiotics. Furthermore, the spectrum of cationic peptides includes some of the most potent toxins (e.g., bee venom and scorpion toxin), so that toxicity will always be a closely observed issue. In addition, being peptides, they are potentially susceptible to host

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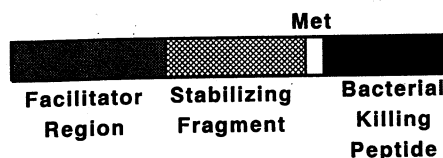


Figure 7 Schematic diagram of a protein A-cationic peptide fusion protein. Met represents the position of the methionine residue used to cleave the peptide with cyanogen bromide.

peptidases and proteases, and innovative approaches will have to be applied to overcome this problem (e.g., the use of D-amino acids). However, activity, toxicity, and pharmacology are issues with every compound used in medicine.

With these reservations, we believe that cationic peptides offer an exciting future. They represent a "natural" solution to infection, since they mimic the anti-infective defense systems of several eukaryotes. Their activities cover a far broader spectrum than do those of other antibiotics. Indeed, they offer the potential for organ-specific therapy directed against the major bacterial and fungal infections of a given body site. In addition, the most active cationic peptides have activities against some of the more refractory antibiotic-resistant pathogens (e.g., *Pseudomonas aeruginosa* and methicillin-resistant *Staphylococcus aureus*) that are equivalent to those observed for the antibiotics tailored for use against those pathogens. Indeed, that they do not seem to induce antibiotic resistance and are effective against most bacteria resistant to conventional antibiotics are important features of cationic peptides. In addition, their potential to act in synergy with conventional antibiotics and to neutralize endotoxin released by these antibiotics makes them an attractive candidate for use in combination therapy.

ACKNOWLEDGMENT

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Rapamycin Compounds

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I. PHARMACEUTICALS

For years, the only antifungal agent cyclosporin. *Tolypocladium inflatum* produces bicyclic molecules produced by ascomycin (2-4). Rapamycin, which causes liver damage, hypotension, and was approved in the late 1980s. It is more potent than cyclosporin. It is present and may be used in the treatment of A and FK506 due to its immunosuppressive properties.

Immunosuppressive drugs include immunophilins (FK506 and rapamycin (FKBP12) (7), which are immunophilins and protein foldases." When FKBP12 is inactive and the production pathways of FK506-FKBP12 complex, namely rapamycin-FKBP12 complex, which is phosphorylated by FKBP12 target), are inhibited.

Inhibition of cytokine presentation; thus, the release of interleukin-2-depending on the concentration of the drug.

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